Note

Methylation analysis of non-glycosidic phosphoric diesters of carbo-hydrates*

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Methylation of synthetic¹ α,β -D-glucose 2- $[\alpha,\beta$ -D-glucose 6-(sodium phosphate)] (1), α,β -D-glucose 3- $[\alpha,\beta$ -D-glucose 6-(sodium phosphate)] (2), and α,β -D-glucose 6- $[\alpha,\beta$ -D-glucose 6-(sodium phosphate)] (3) according to Hakomori² was not satisfactory. After borodeuteride-reduction of 1-3 (yielding 4, 6, and 8, respectively) followed by methylation², the formation of 5, 7, and 9, respectively, was expected. However, g.l.c.-m.s. revealed that large proportions of each of the latter phosphoric triesters had been cleaved, mainly into the corresponding glucose dimethyl phosphates (molar ratios of di- to mono-glucose triesters, 1:2-1:10) together with smaller amounts of other products, including anhydro forms (data not shown).

Esterification of 4, 6, and 8 with diazomethane in ether followed by methylation⁴ gave the phosphoric triesters 5, 7, and 9, which were identified by g.l.c. and g.l.c.-m.s. (Table I). Only small amounts of mono-glucose phosphoric triesters were formed. Cleavage of 5 with LiAlH₄ in ether⁵ followed by acetylation gave 2-O-acetylpenta-O-methylhexitol-1-d (10) and 6-O-acetylpenta-O-methylhexitol-1-d (11). Likewise, 7 gave 3-O-acetylpenta-O-methylhexitol-1-d (12) and 11, and 9 gave 11. Compounds 10-12 were identified by g.l.c.-m.s. The ratio of 10 and 12 to 11 was \sim 1:1. Using splitless injection, 150 μ g were sufficient for the derivatisation procedure described.

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TABLE 1
G.L.C. AND E.I.-M.S. DATA FOR 5, 7, AND 9

Compound	Mol. wt.ª	Retention time ^b	m/z (% of base peak)
5	582	3.66	45 (22.2), 46 (12.2), 60 (11.1), 71 (24.4), 72 (18.9), 75 (20.0), 88 (20.0), 89 (24.4), 90 (22.2), 101 (100), 102 (32.2), 116 (20.0), 127 (8.9), 146 (28.9), 172 (26.7), 236 (48.9)
7	582	3.65	45 (11.1), 46 (8.9), 60 (10.0), 71 (13.3), 72 (7.8), 75 (15.6), 88 (10.0), 89 (13.3), 90 (17.8), 101 (100), 102 (14.4), 116 (26.7), 146 (14.4), 172 (30.0), 236 (27.8)
9	582	4.22	45 (17.8), 46 (13.3), 60 (13.3), 71 (40.0), 72 (16.7), 75 (24.4), 88 (28.9), 89 (15.6), 90 (27.8), 101 (100), 102 (25.6), 113 (20.0), 116 (15.5), 127 (10.0), 146 (17.8), 172 (12.2), 236 (12.2)

^aDetermined by c.i. (ammonia)-m.s. on the basis of peaks at m/z for $(M+1)^+$ and $(M+18)^+$. ^bRelative to that of α -D-glucose penta-acetate; see Experimental.

EXPERIMENTAL

Compounds 1–3 were synthesised as described¹ and conventional reduction with NaB²H₄ afforded 4, 6, and 8, respectively, separate solutions (1–2 mg in 0.1 mL) of which in methanol were treated with ethereal diazomethane (1 mL) for 30 min at room temperature. After concentration, each residue was dried over P_2O_5 and dissolved in dry Me₂SO (1 mL), and finely powdered dry NaOH (~25 mg) and MeI (1 mL) were added⁴. Each mixture was kept for 15 min at room temperature,

then diluted with water (2 mL) and extracted twice with CH_2Cl_2 . The combined extracts were concentrated, and a solution of the residue in Me_2SO (0.2 mL) was diluted with water (0.2 mL) and purified on a C_{18} cartridge³ (SEP-PAK) by washing in sequence with water and acetonitrile-water (1:10) followed by elution with acetonitrile-water (1:5). A solution of each methylated phosphoric diester in ether (0.5 mL) was treated (30 min, room temperature)⁵ with LiAlH₄ (~5 mg). After evaporation of the solvents, the residue was kept under nitrogen and water was added. The resulting salts were removed by centrifugation, the supernatant solution was concentrated, and the residue was acetylated conventionally (acetic anhydride-pyridine).

G.l.c. was performed on a Varian 3700 gas chromatograph equipped with a flame-ionisation detector and a fused-silica capillary column (25 m \times 0.32 mm i.d.) with chemically bonded SE-54 (0.2 μ m) (Weeke, Mühlheim), H₂ as the carrier gas (1.5 bar), and temperature programmes of 160° for 3 min, 5°/min \rightarrow 300° (for 5, 7, and 9), and 130° (for partially methylated hexitols).

G.l.c.-m.s. was performed on a Hewlett-Packard 5985 instrument equipped with an SE-54 column and an HP-1000 data system. E.i.-mass spectra were recorded at 70 eV and c.i.-mass spectra were obtained with ammonia as reactant gas. The ion-source temperature was 200°.

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